

# Equations of State

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# Outline

- What is the equation of state and why should we care?
- Theoretical basis
- Complete and incomplete EOS
- Predicting the EOS
- Experimental measurements
- Corrections, limits, and other issues

# Nomenclature for thermodynamics

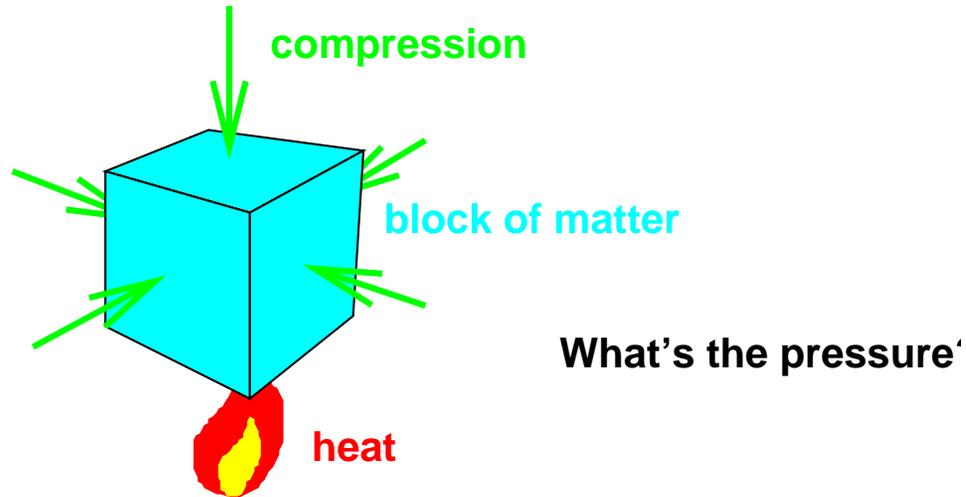
Mass density  $\rho$  and specific volume  $v$ :  $\rho = 1/v$ .

<b>intensive</b>	<b>extensive</b>
$p$ pressure	$e$ internal energy
$T$ temperature	$s$ entropy

Extensive quantities: can integrate over system to find total.

Convention here: use specific quantity (per mass).

# What is the equation of state?



**Material-dependent** property, relating **thermodynamic potential** and its **natural parameters**, e.g.  $de = Tds - pdv$ .  
EOS is the relation  $e(s, v)$  for a material.

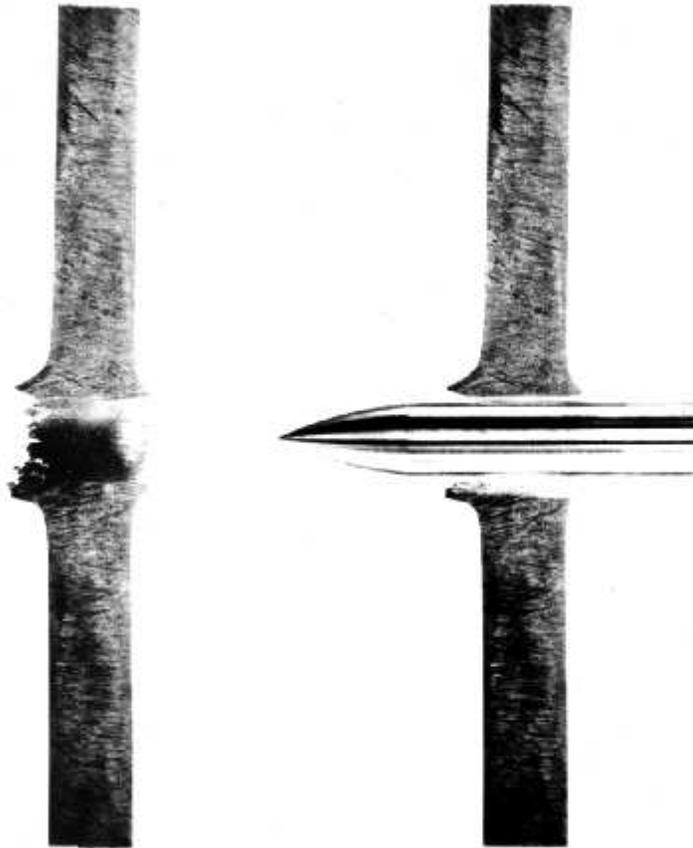
Derivatives and other functions of EOS give other quantities:  
 $p$ ,  $T$ , sound speed  $c^2$ , heat capacity  $c_v$ , ...

Perfect gas EOS:

$$p = nk_B T \quad p = (\gamma - 1)\rho e$$

# Uses of the equation of state

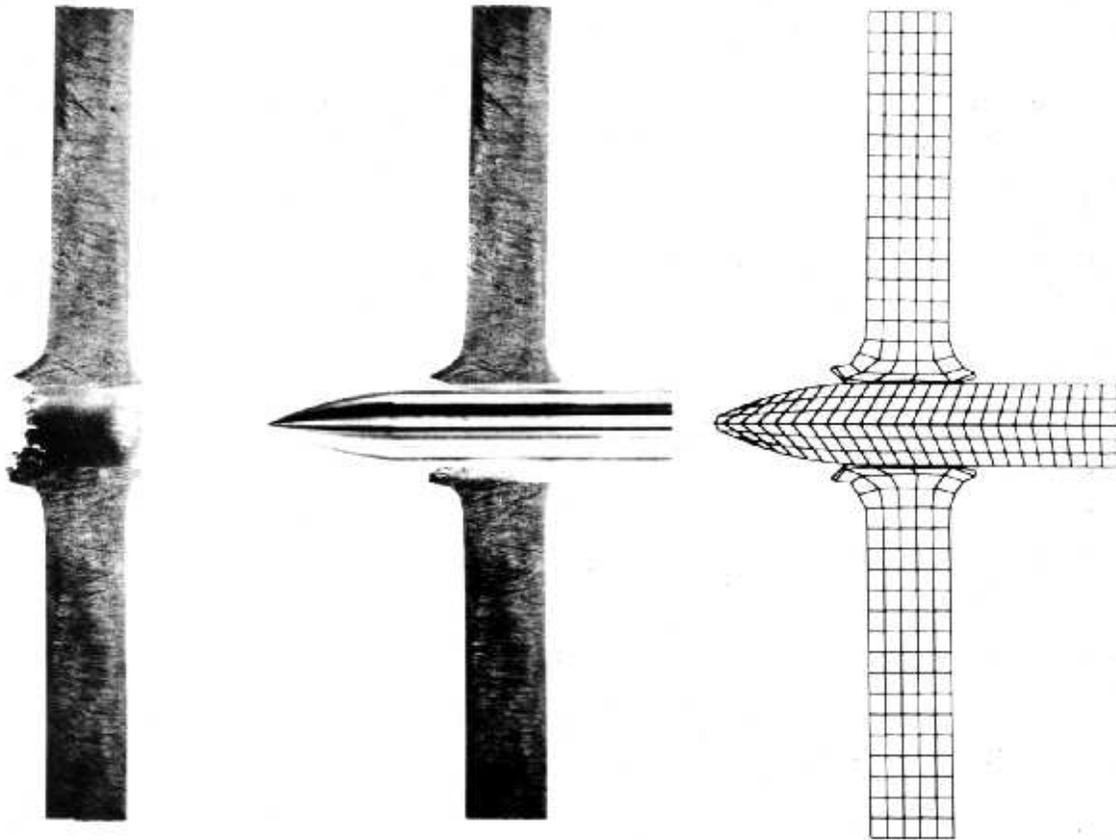
Hydrocode / continuum mechanics simulations of impact-type problems:



Cross-section of armor after perforation by projectile.

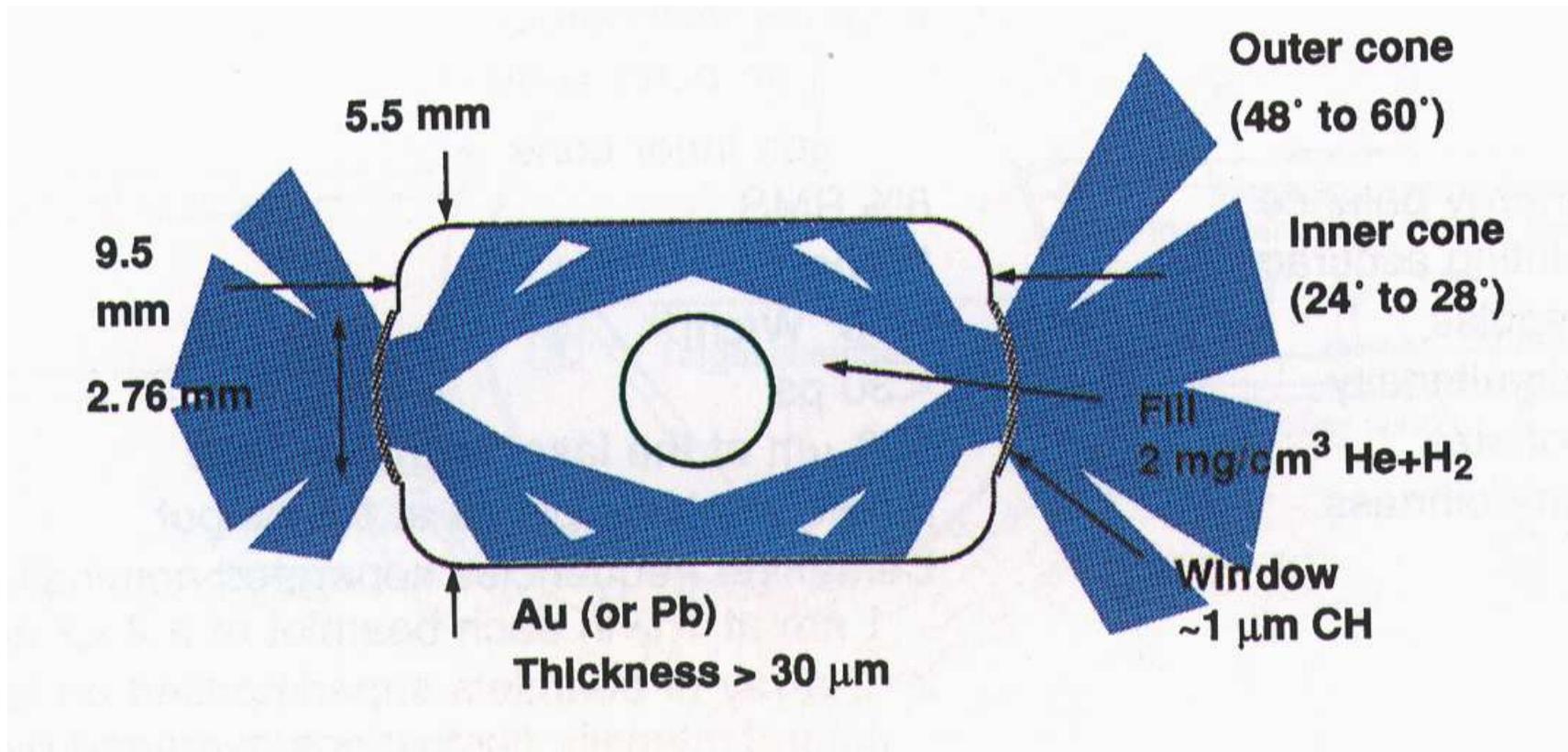
# Uses of the equation of state

Divide components into small cells, assume conditions are spatially uniform in each. Need material properties for simulations.



# Uses of the equation of state

Radiation-driven implosions, e.g. NIF hohlraum-driven fusion capsule:



Source: J. Lindl, "Inertial confinement fusion," Springer (1998).

# Simulation of dynamic loading problems

Initial value problem: given  $\{\rho, \vec{u}, e\}(\vec{r}, t_0)$  over some region  $\{\vec{r}\} \in \mathcal{R}$ , what is  $\{\rho, \vec{u}, e\}(\vec{r}, t > t_0)$ ?

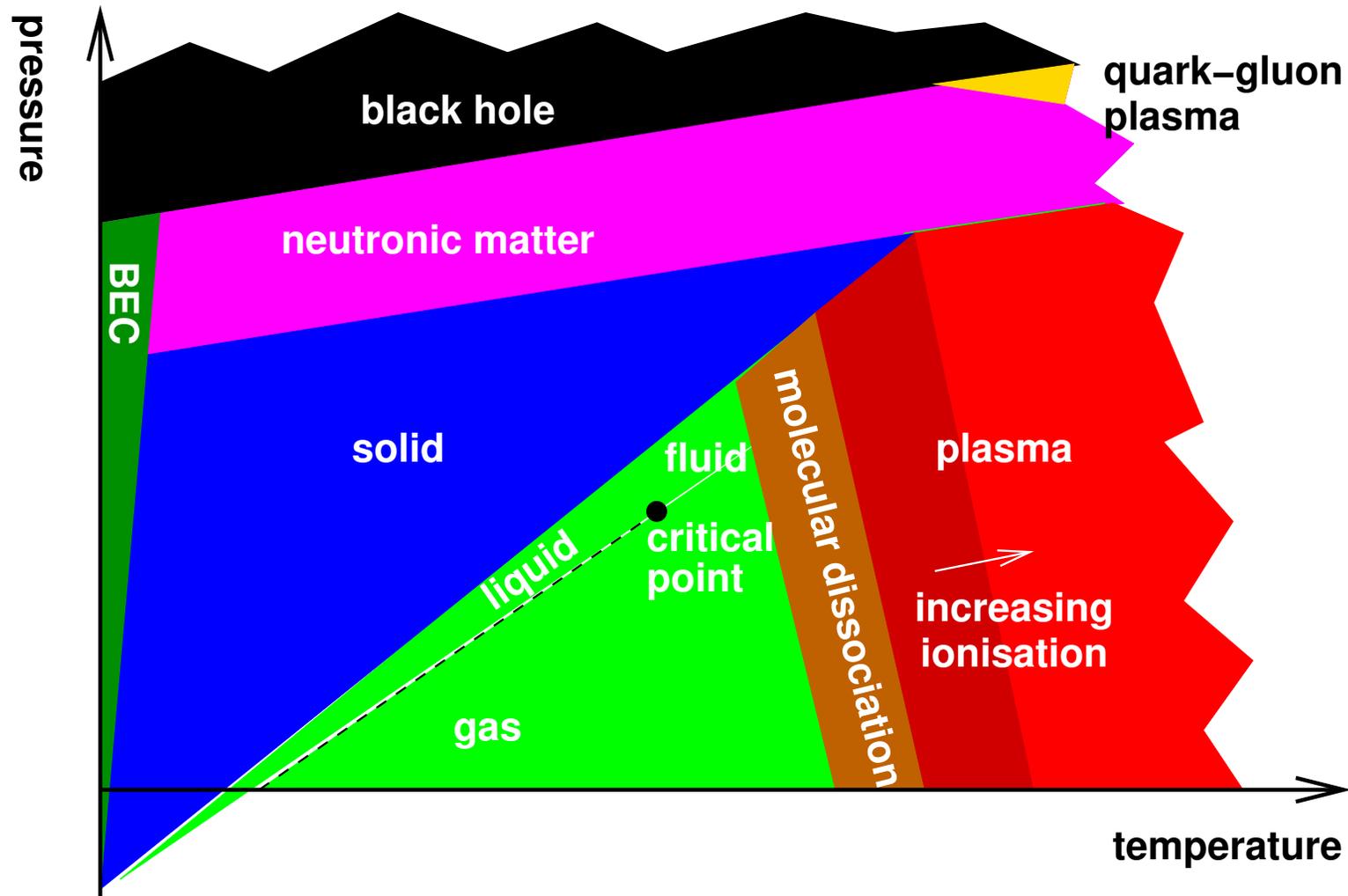
Continuum equations (Lagrangian, neglecting heat conduction):

$$\begin{aligned}\frac{\partial \rho(\vec{r}, t)}{\partial t} &= -\rho(\vec{r}, t) \operatorname{div} \vec{u}(\vec{r}, t) \\ \frac{\partial \vec{u}(\vec{r}, t)}{\partial t} &= -\frac{1}{\rho(\vec{r}, t)} \operatorname{grad} p(\vec{r}, t) \\ \frac{\partial e(\vec{r}, t)}{\partial t} &= -p(\vec{r}, t) \operatorname{div} \vec{u}(\vec{r}, t)\end{aligned}$$

Boundary conditions  $p(\vec{r}, t)$  and/or  $\vec{u}(\vec{r}, t)$  for  $\{\vec{r}\} \in d\mathcal{R}$ .

Use EOS  $p(\rho, e)$  – derived from  $e(s, v)$  – to complete equations and allow integration to proceed.

# Structure of the equation of state



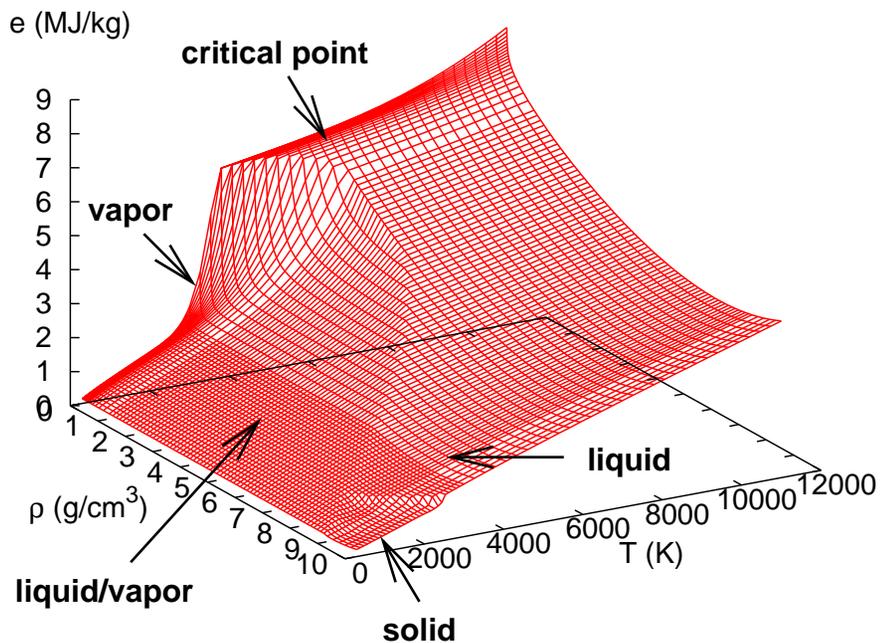
Details depend on the material composition.

# 'Moderate' pressure and temperature

Geophysics, hypervelocity impact, terrestrial explosions, ICF, ...

–GPa ( $p_{\text{spall}}$ ) <  $p$  < PPa (Gbar);  $\sim 10 \text{ K} < T < \text{MeV}$ ;  $\text{ps} < t < \text{Gyr}$

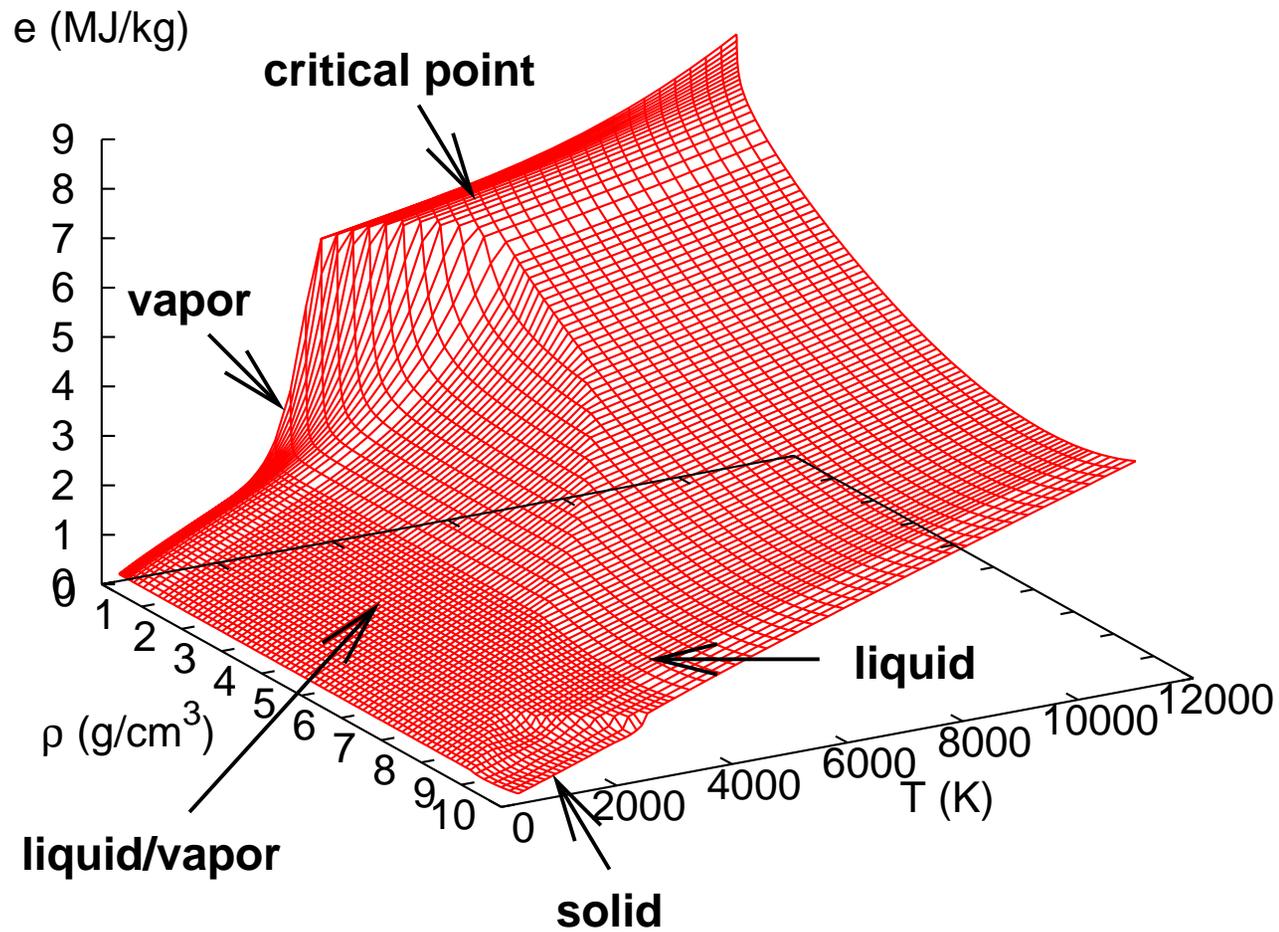
e.g. Cu:



$$e(\rho, T) = e_{\text{cold}}(\rho) + e_{\text{ion-thermal}}(\rho, T) + e_{\text{electron-thermal}}(\rho, T)$$

# Cu phase diagram

SESAME #4:



## Energy, EOS, and phase diagram

Given  $e(\rho, T)$  – for a given atomic arrangement – can construct EOS using 2nd law of thermodynamics:

$$de = Tds - pdv \quad \Rightarrow \quad s(v, T) = s(v, 0) + \int_0^T \frac{dT'}{T'} \frac{\partial e(v, T')}{\partial T'}$$

from which calculate  $f(\rho, T) = e(\rho, T) - Ts(\rho, T) \Rightarrow$  EOS for this phase.

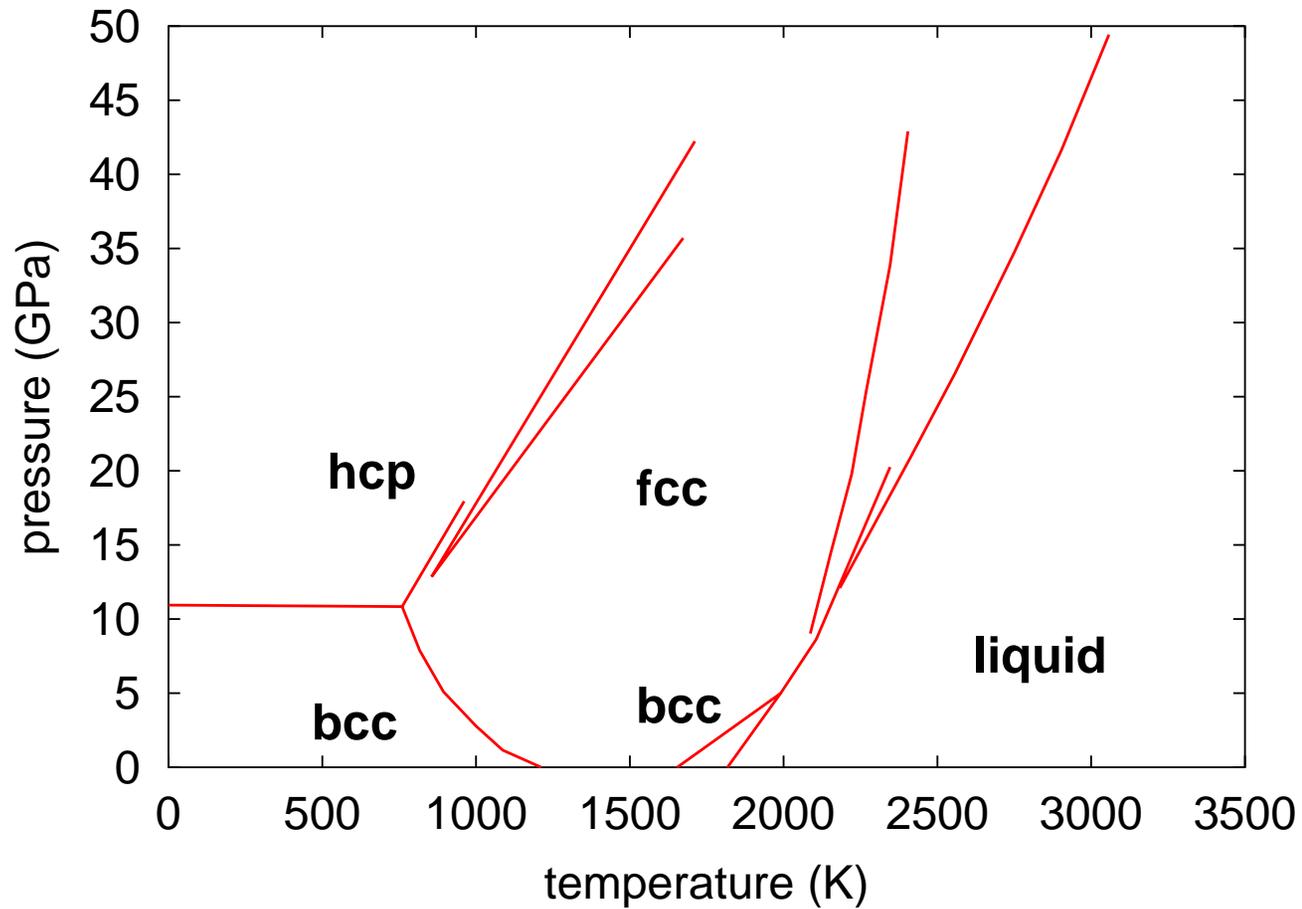
$s(v, 0)$ : ‘configurational entropy’ – may vary between phases.

At any  $p, T$ , equilibrium phase has lowest Gibbs free energy

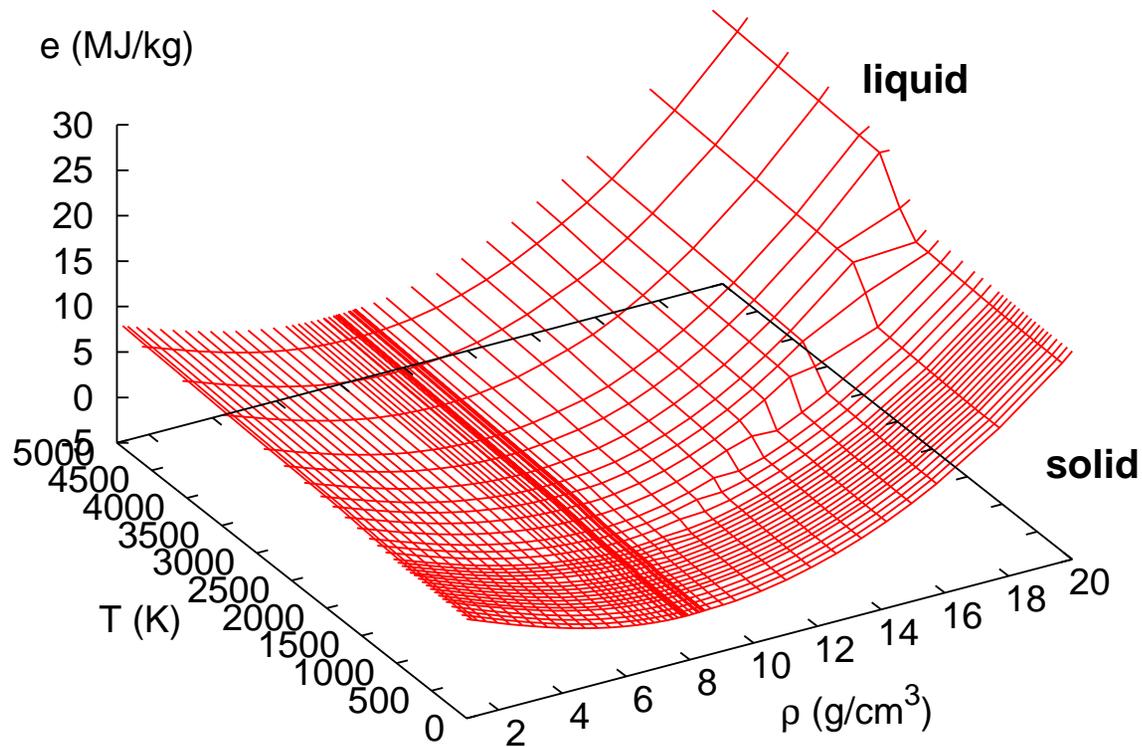
$$g = e - Ts + pv \quad (\text{natural parameters } g(T, p) \text{ as } dg = -sdT + vdp)$$

# Phase diagram: Fe

Different crystal structures occur in the solid state:

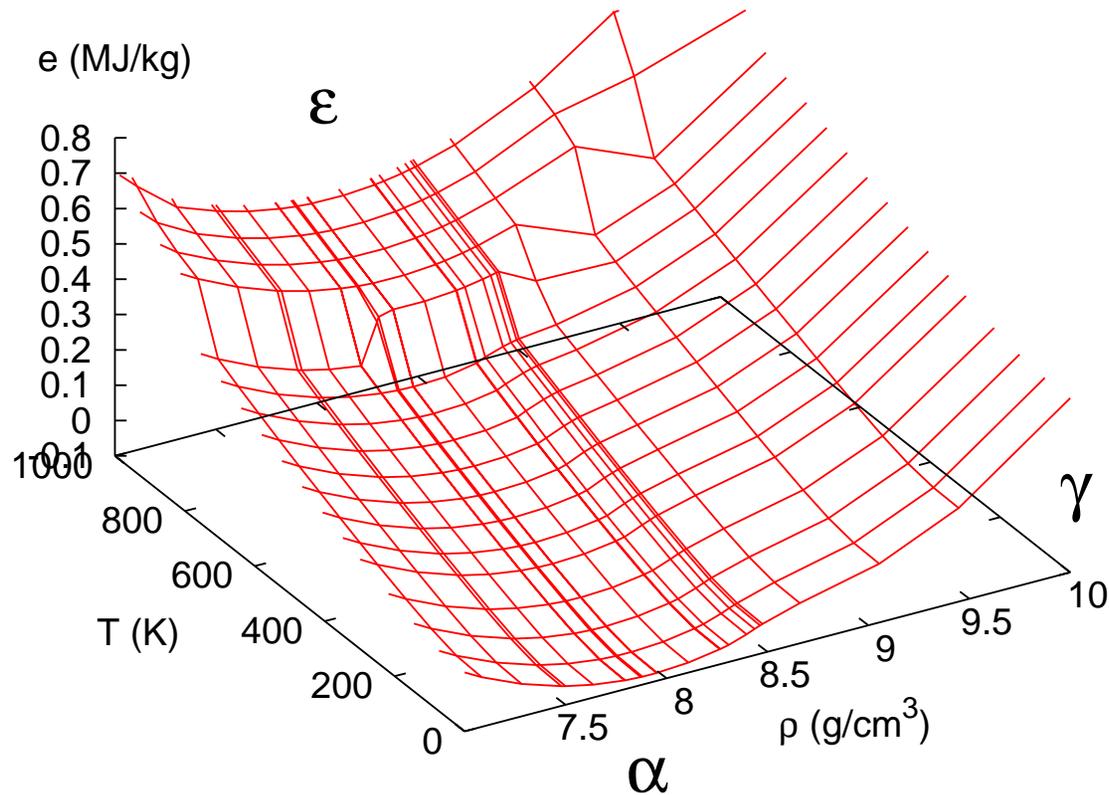


# Multiphase EOS: Fe



Source: J.T. Gammel, T-1.

# Multiphase EOS: Fe



Source: J.T. Gammel, T-1.

# Thermodynamic completeness

Complete EOS: thermodynamic potential expressed in natural parameters

$$e(s, v), f(T, v), g(T, p), h(s, p)$$

Can derive any thermodynamic parameter from a complete EOS.

Continuum mechanics:

$$\partial\rho/\partial t = -\rho\text{div } \vec{u}, \quad \partial\vec{u}/\partial t = -(1/\rho)\text{grad } p, \quad \partial e/\partial t = -p\text{div } \vec{u}$$

only require  $p(\rho, e)$ .

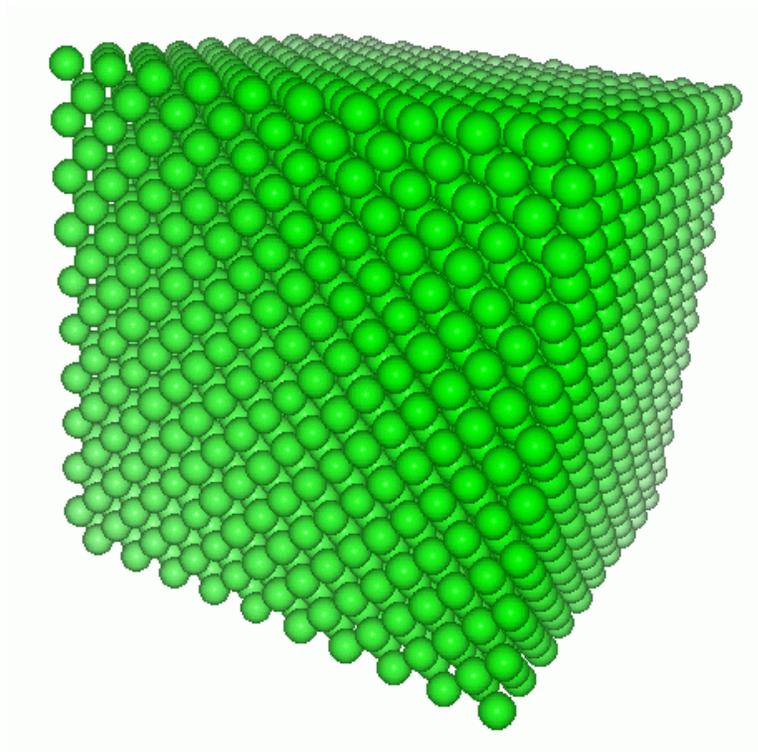
Also: easiest to deduce  $p(\rho, e)$  from mechanical measurements.

Incomplete EOS:  $p(\rho, e)$  with no information about  $T$ .

‘SESAME’ EOS:  $e(\rho, T)$  and  $p(\rho, T)$ : a complete EOS – but may be inconsistent.

# Predicting the EOS

$$e(\rho, T) = e_{\text{cold}}(\rho) + e_{\text{ion-thermal}}(\rho, T) + e_{\text{electron-thermal}}(\rho, T)$$



– solve for electron states for a given set of ion positions (Dirac or Schrödinger equation).

Electrons are indistinguishable fermions: quantum many-body problem...

# Quantum many-body problem

Ground state of collective wavefunction  $\Psi$ :  $\hat{H}\Psi = E_0\Psi$ .

Numerical quantum mechanics: based on single-particle wavefunctions  $\{\psi_i\}$ . Fermions: antisymmetric with respect to particle exchange (Pauli exclusion principle), e.g. for 2-particle wavefunctions  $\Psi(1, 2) = -\Psi(2, 1)$ , can be satisfied if

$$\Psi(1, 2) = \frac{1}{\sqrt{2}}[\psi_1(1)\psi_2(2) - \psi_2(1)\psi_1(2)].$$

More generally,

$$\Psi(1, 2, \dots, n) = \sum_{\chi} \epsilon(\chi) \psi_{\chi(i)}(i) = \det \psi_i(j)$$

where  $\epsilon$  is a permutation operator – ‘Slater determinant’ approach; in practice prohibitive in computer time.

Differences in the phase of the wavefunctions  $\Rightarrow$  correlation effects: another complication.

# Local density approximation

Assume can approximate exchange and correlation by modifying the potential energy contribution in the Hamiltonian to include functions of the local electron density

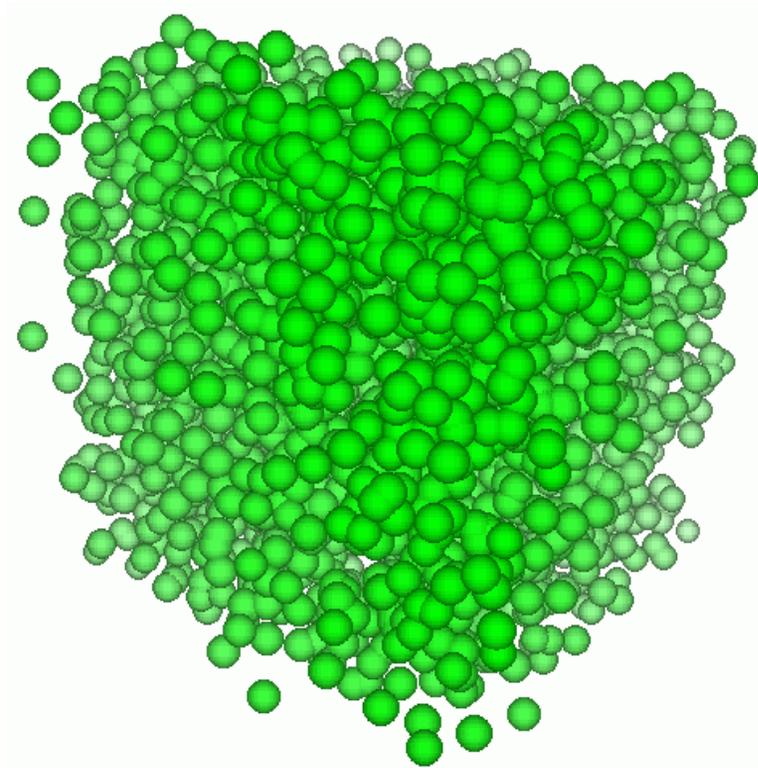
$$n(\vec{r}) = \sum_i \psi_i^\dagger(\vec{r})\psi_i(\vec{r}).$$

These functions are then calibrated against detailed exchange and correlation calculations for simple systems e.g. uniform electron gas.

Typical accuracy: a few percent in density at  $p = 0, T = 0$ .

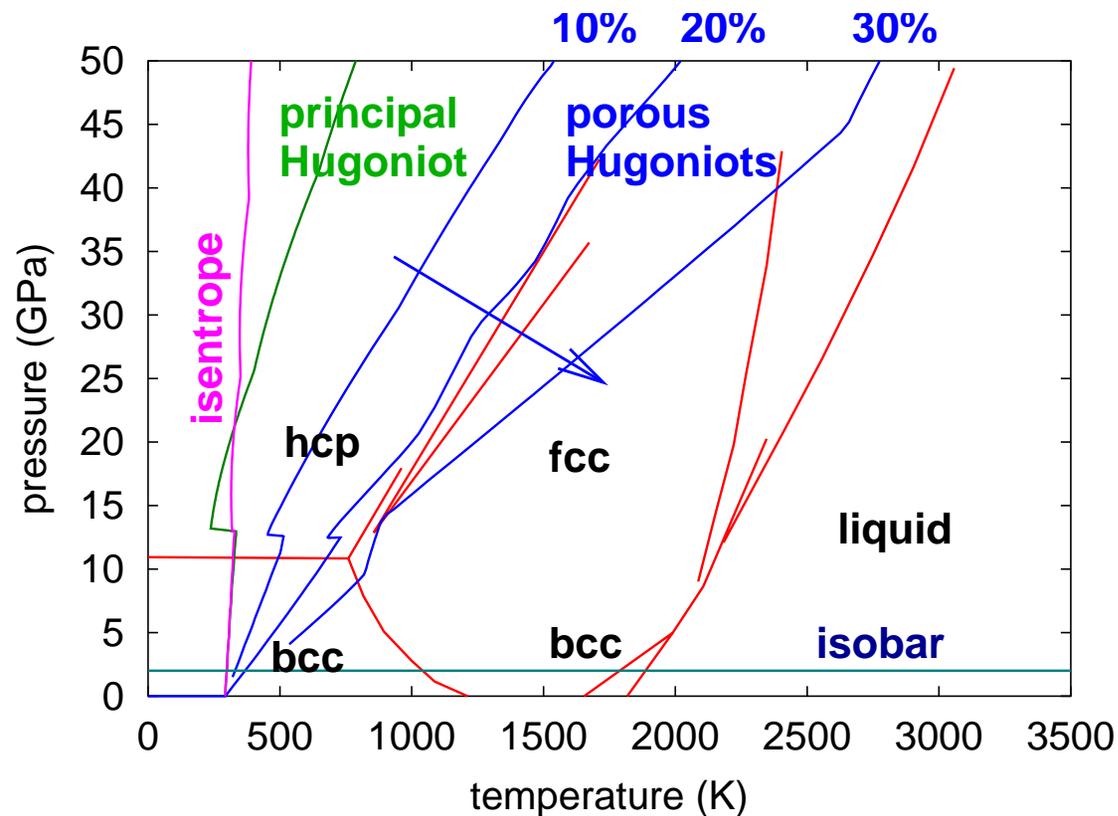
# High temperature EOS

Below  $\sim 1$  eV: ion-thermal from phonons (Bose-Einstein) and electron-thermal from band structure around Fermi surface (Fermi-Dirac).



Different ion positions (may be small effect: dominated by  $\rho$ ); excited electrons / ionization; band structure consistent with  $T$ . Treats plasma seamlessly. Spherical atom model.

# Measuring the EOS

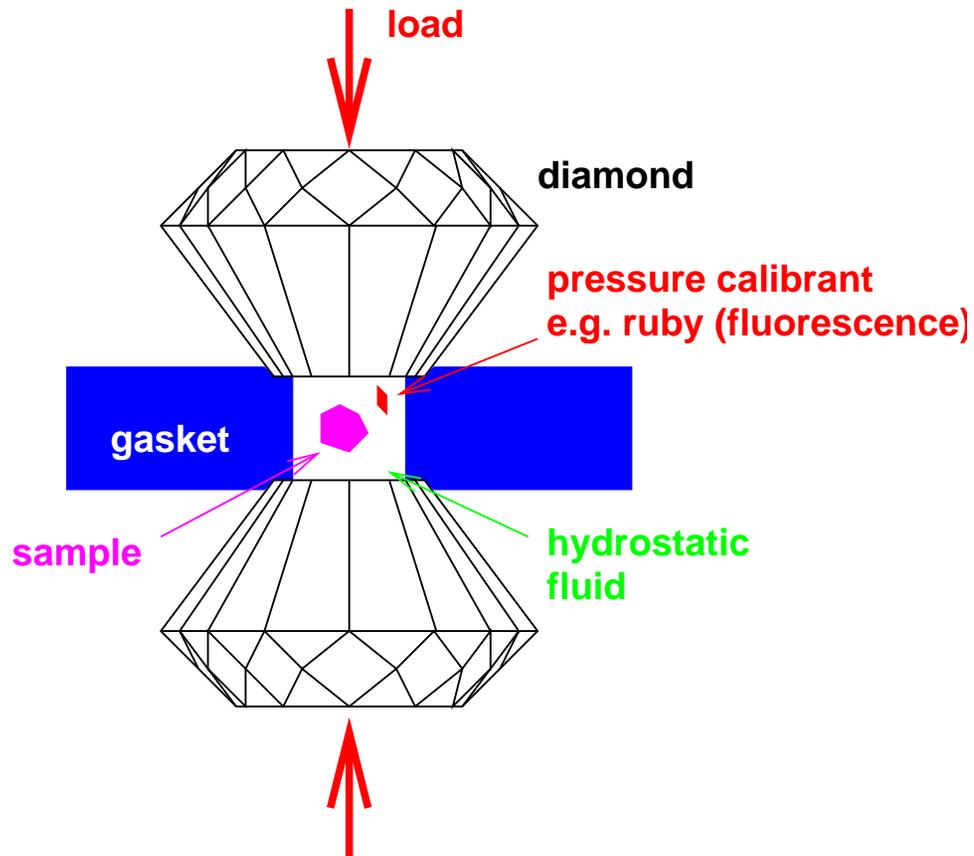


**Isobaric expansion:** exploding wire in pressure vessel.

**Isotherm:** static presses (special case).

**Hugoniot:** locus of states accessible by a single shock wave (can vary its strength).

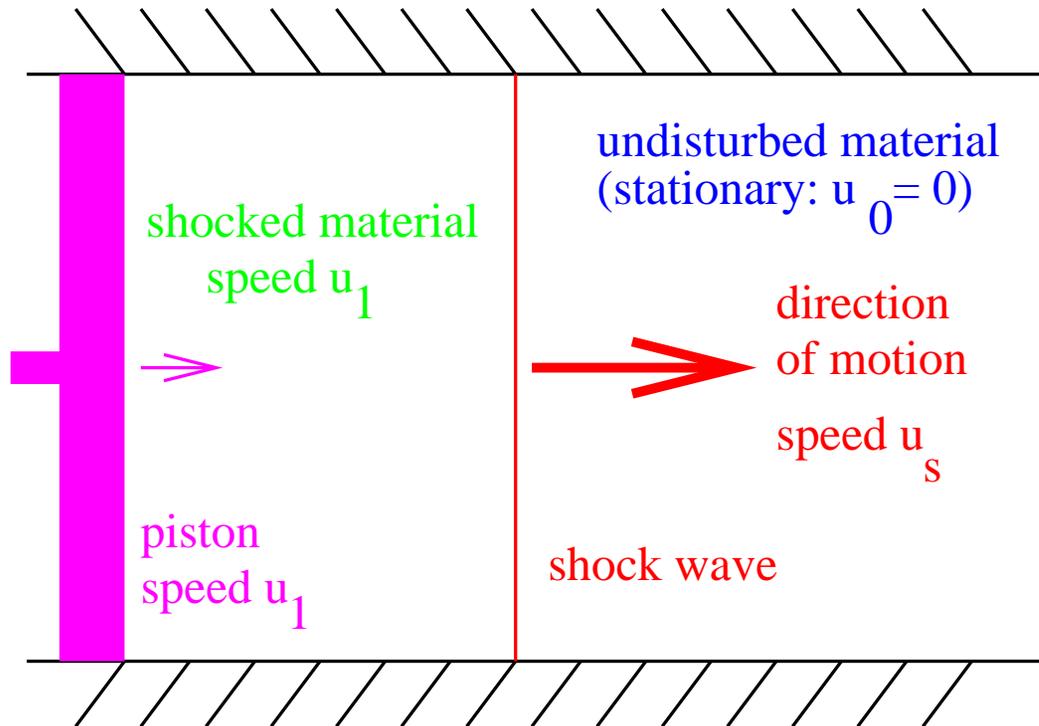
# Static high pressure



Apply load at known temperature; measure pressure, mass density, structure, etc. Diagnostics: x-ray diffraction, optical properties.

$p$  to  $\sim 400$  GPa, steady  $T$  to 2500 K, higher by laser heating.

# Steady shock waves



Rankine-Hugoniot relations (conservation laws):

$$u_s^2 = v_0^2 \frac{p - p_0}{v_0 - v}, \quad u_p = \sqrt{[(p - p_0)(v_0 - v)]}, \quad e = e_0 + \frac{1}{2}(p + p_0)(v_0 - v)$$

– 3 equations, 5 unknowns  $\Rightarrow$  measure 2 quantities to determine state on shock Hugoniot.

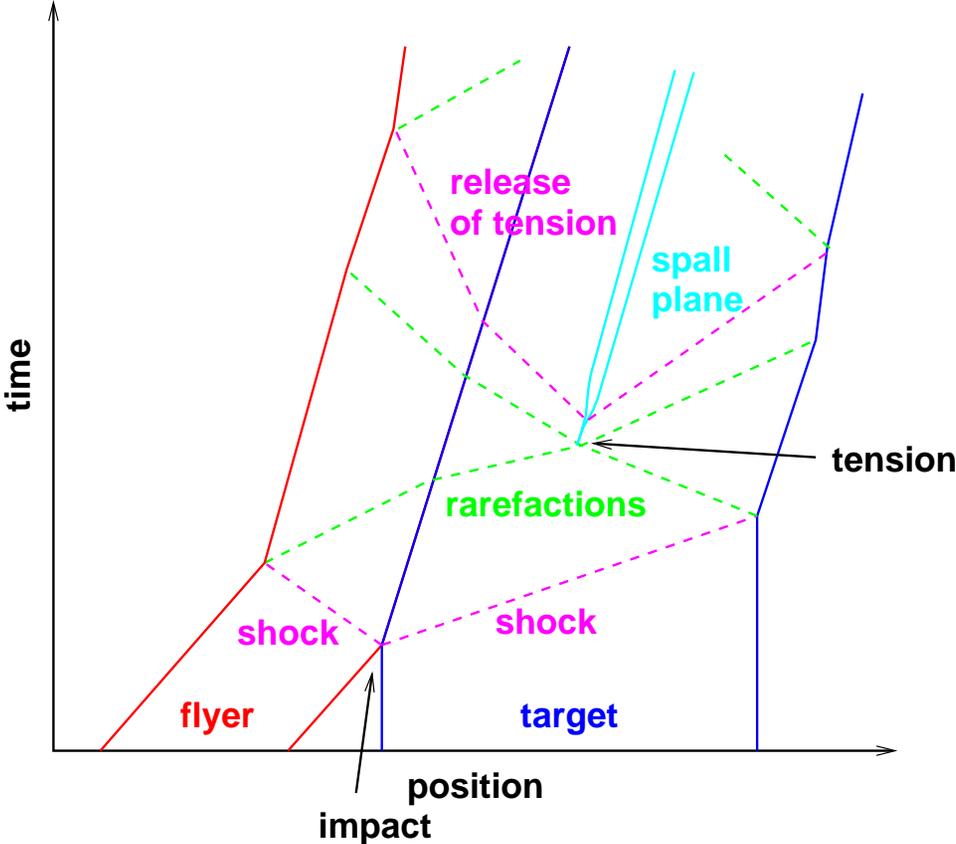
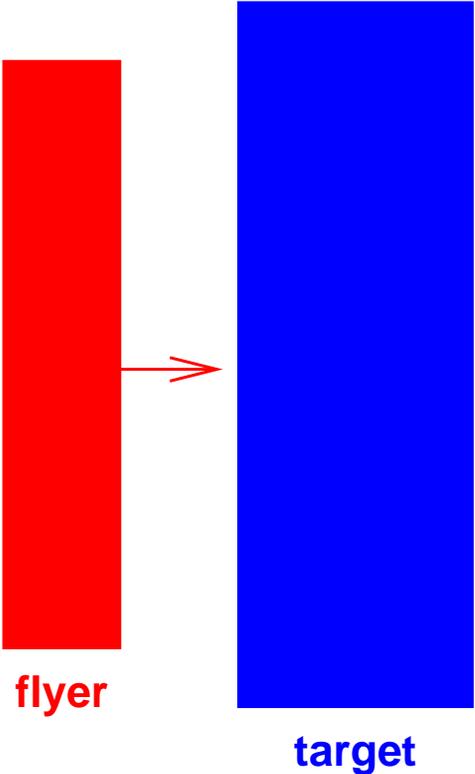
# Shock wave experiments

– ways to launch and measure a steady shock:

- Impact experiments (gas gun, powder gun, electromagnetic flyer, explosively-driven flyer, laser flyer)
- Detonation-driven shock
- Radiation-driven shock (nuclear explosion, laser hohlraum)
- Laser ablation

Aspect ratio: thin samples to preserve 1D region in center.

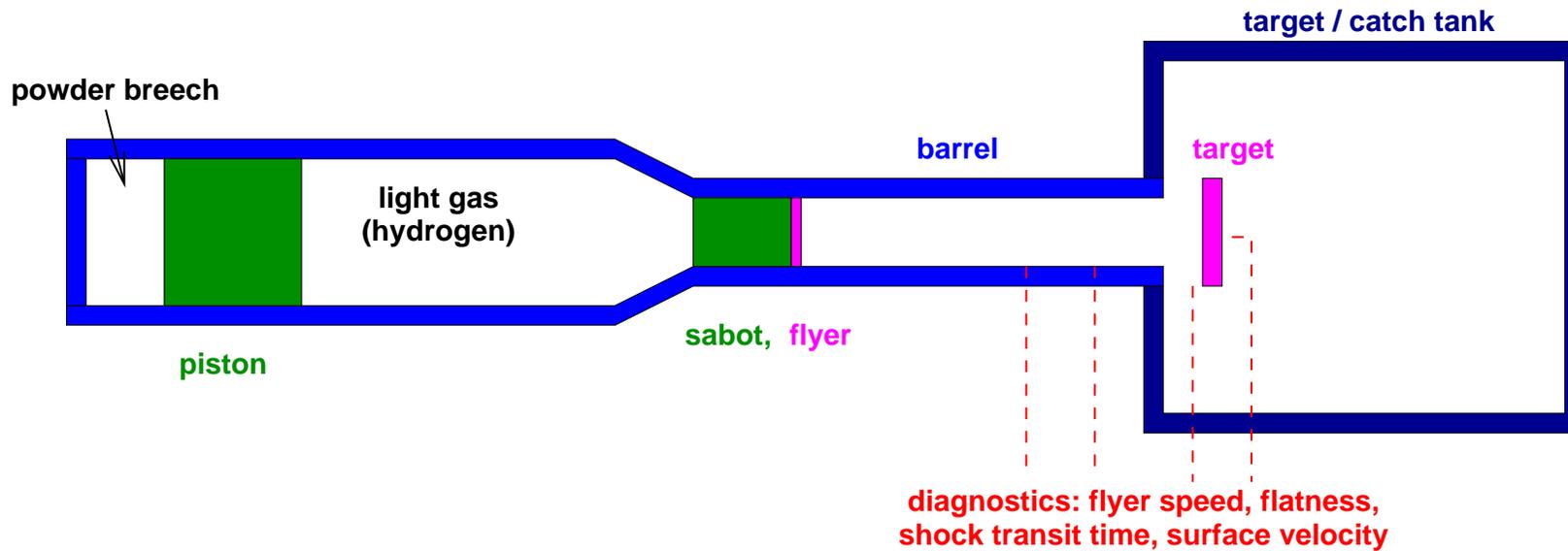
# Impact experiments



“Classical”: same material for both; measure  $u_{\text{flyer}} = 2u_p$  and  $u_s$  from transit time.

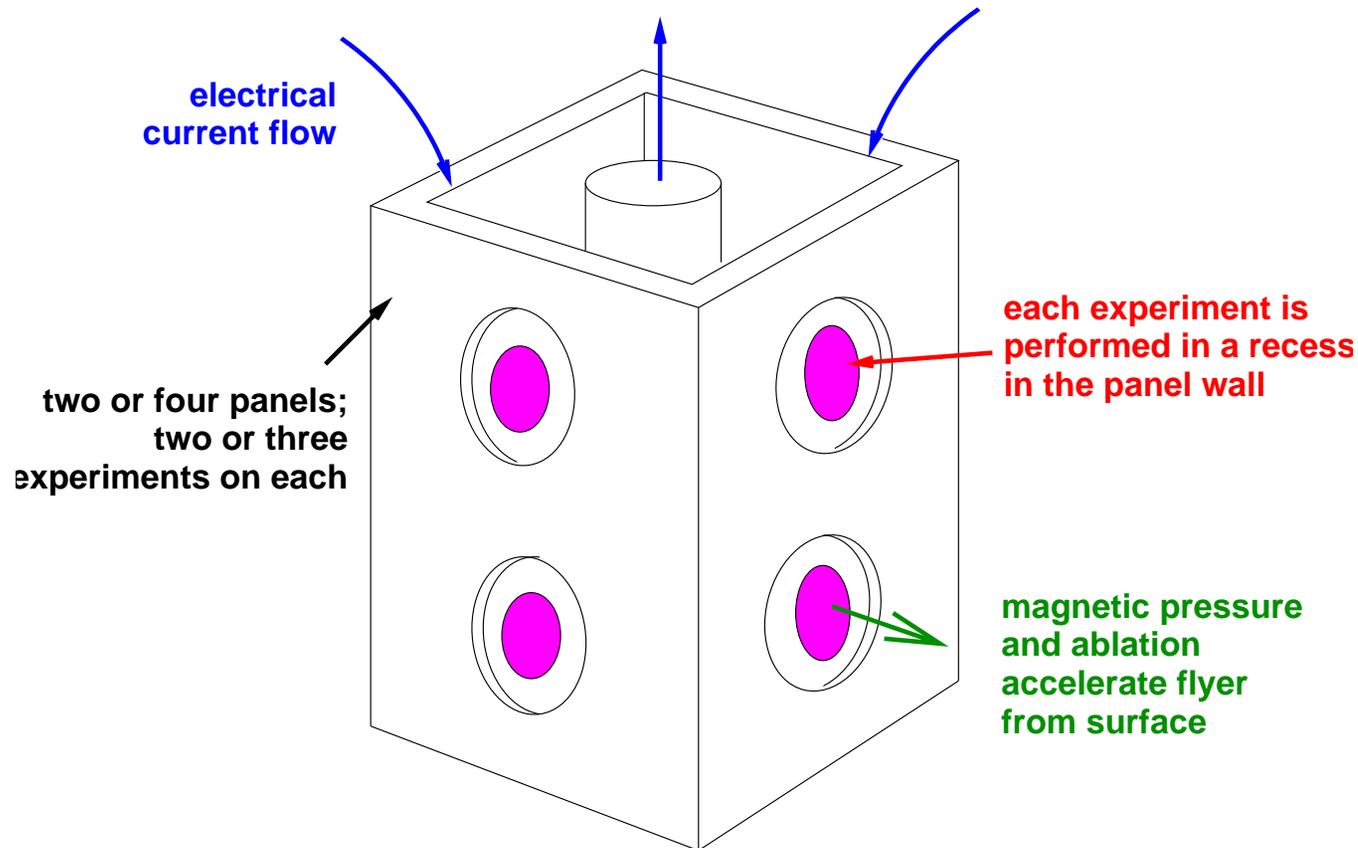
# Gas guns

e.g. two-stage light gas gun:

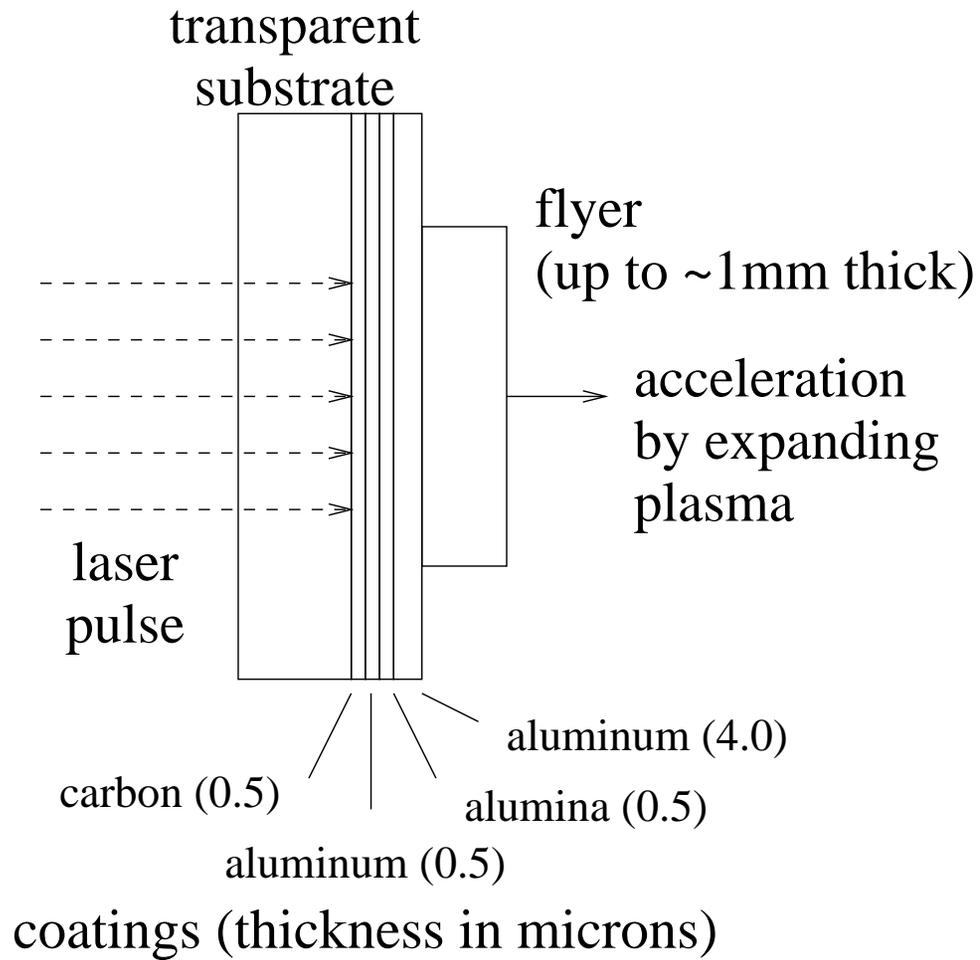


# Electromagnetic launcher

e.g. Z pulsed power machine:



# Laser flyer



# Laser flyer

10 mm

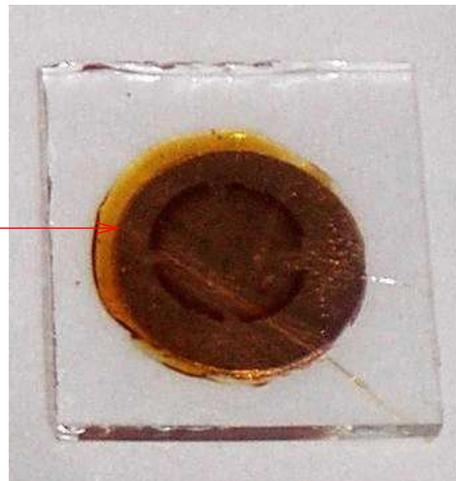


substrate  
(soda-lime glass)

spacer ring  
(dried coffee)

flyer  
(copper)

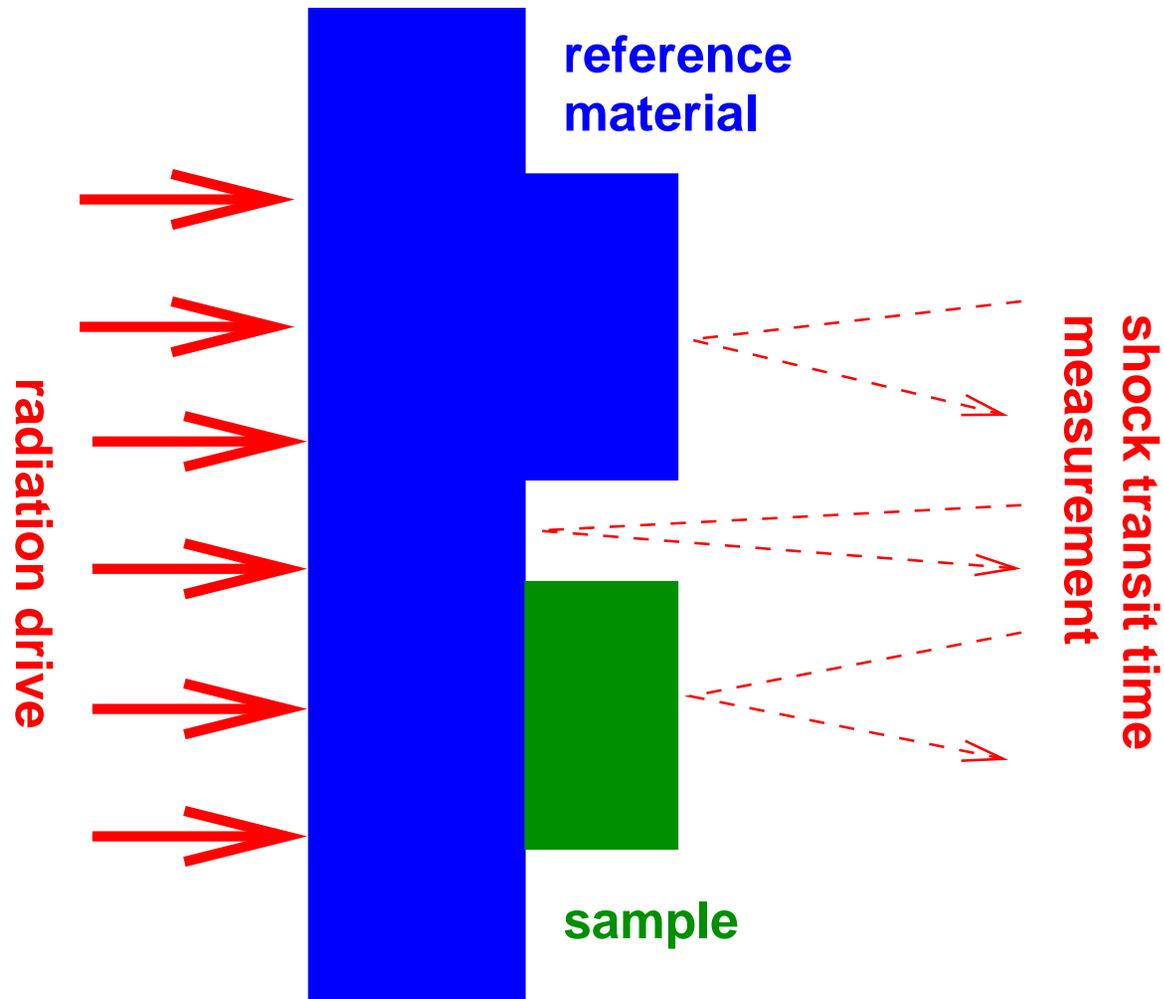
working fluid  
(molasses)



assembly  
(view through substrate)

# Radiative ablation

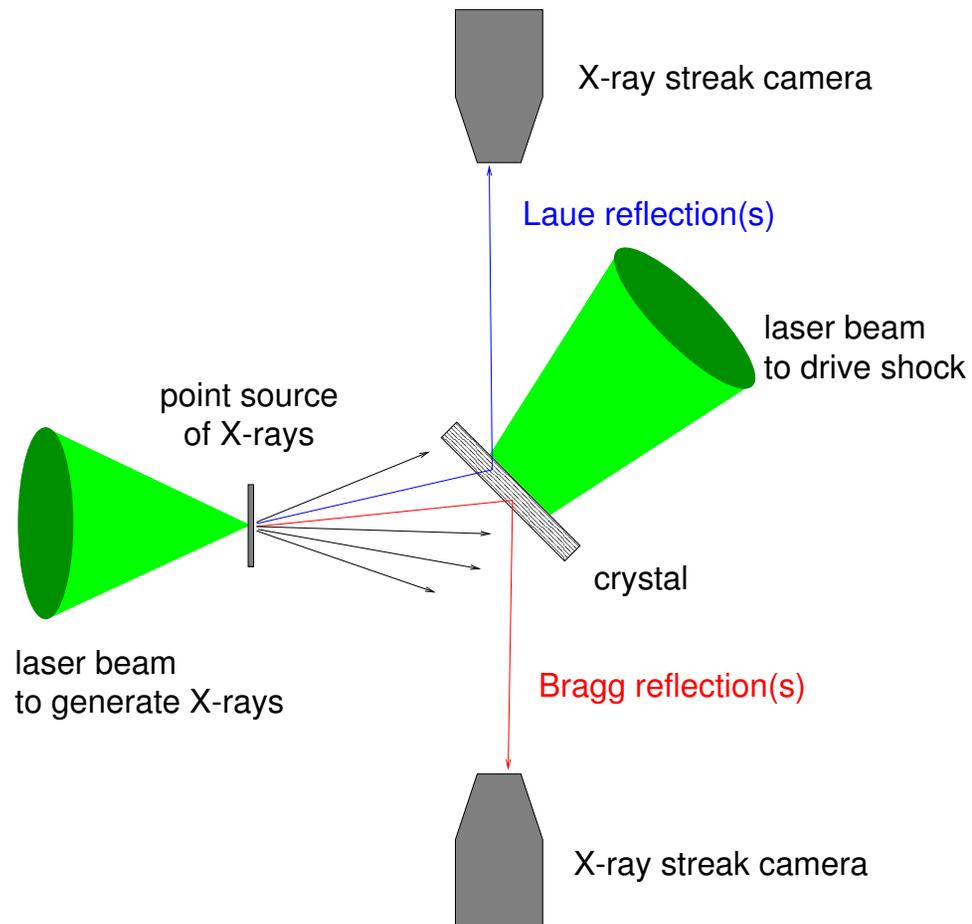
Material ablation drives shock wave by reaction and plasma pressure.



Radiation source: nuclear explosion, laser hohlraum, laser ablation.

# Other shock diagnostics

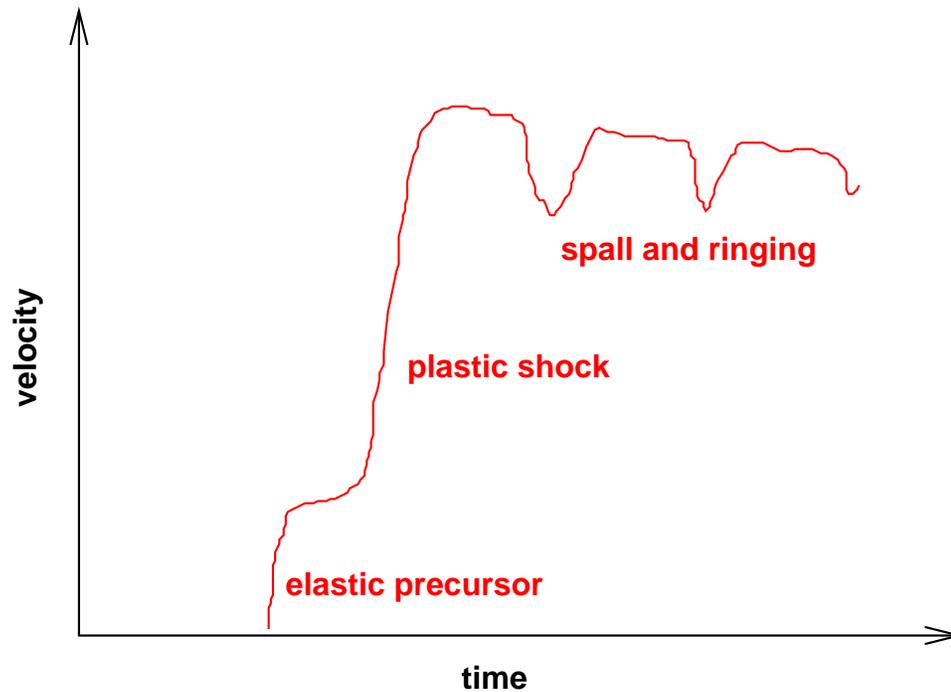
surface velocimetry, radiography, transient x-ray diffraction



# Surface velocimetry

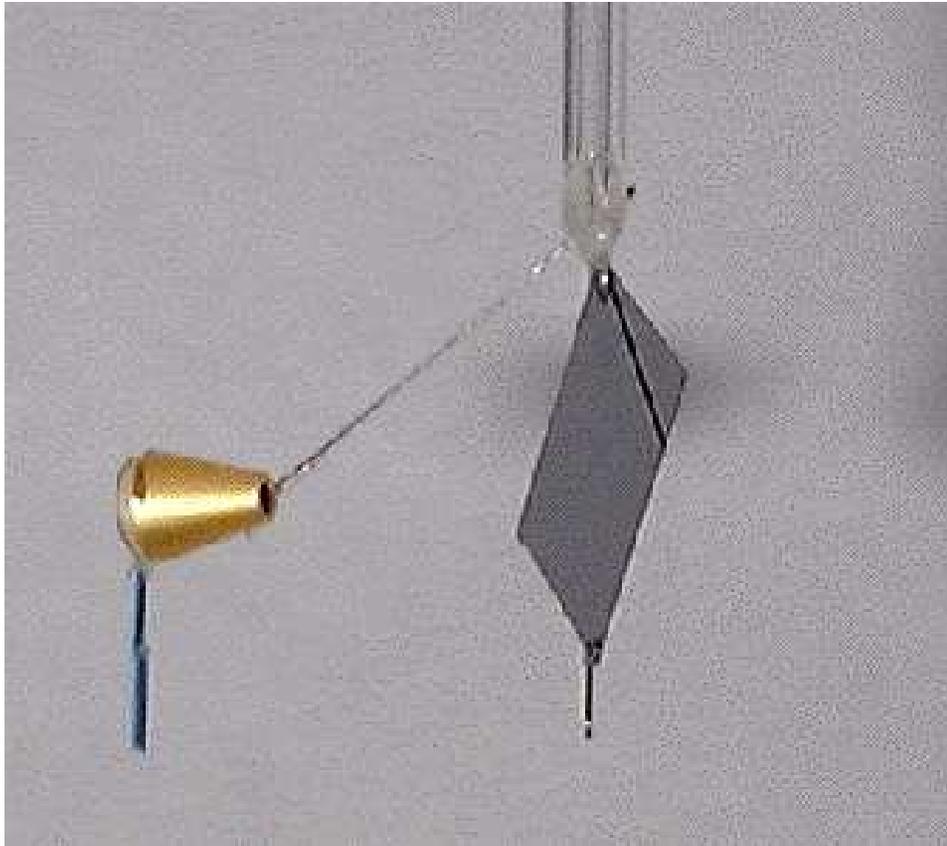
Doppler shift of reflected laser light.

Example velocity history as shock reaches surface:

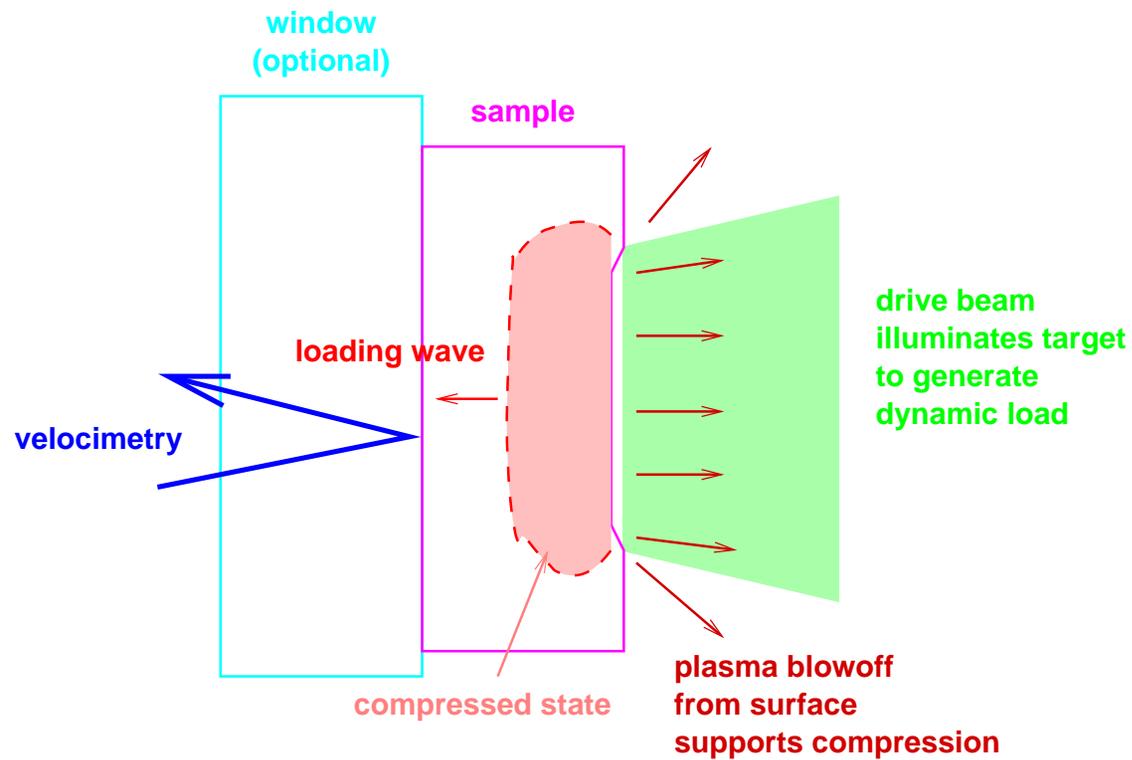


# Transient x-ray diffraction

Si crystal, (100) orientation,  $40\ \mu\text{m}$  thick by 10 mm across:

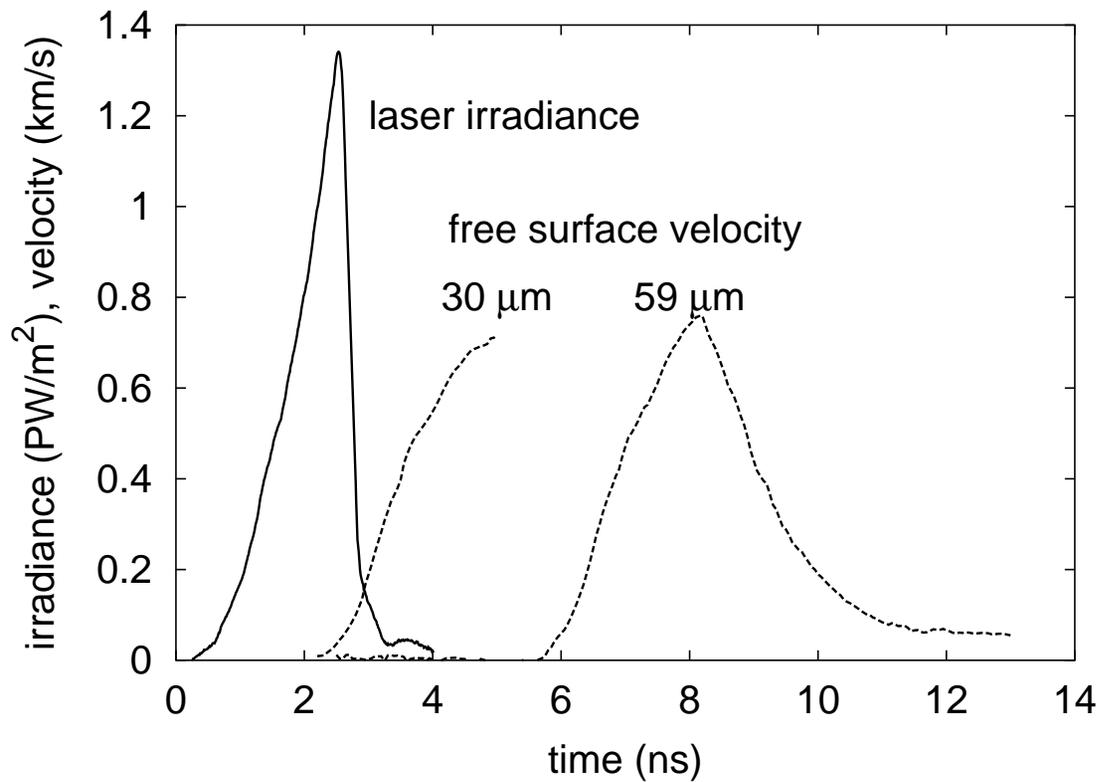


# Quasi-isentropic compression



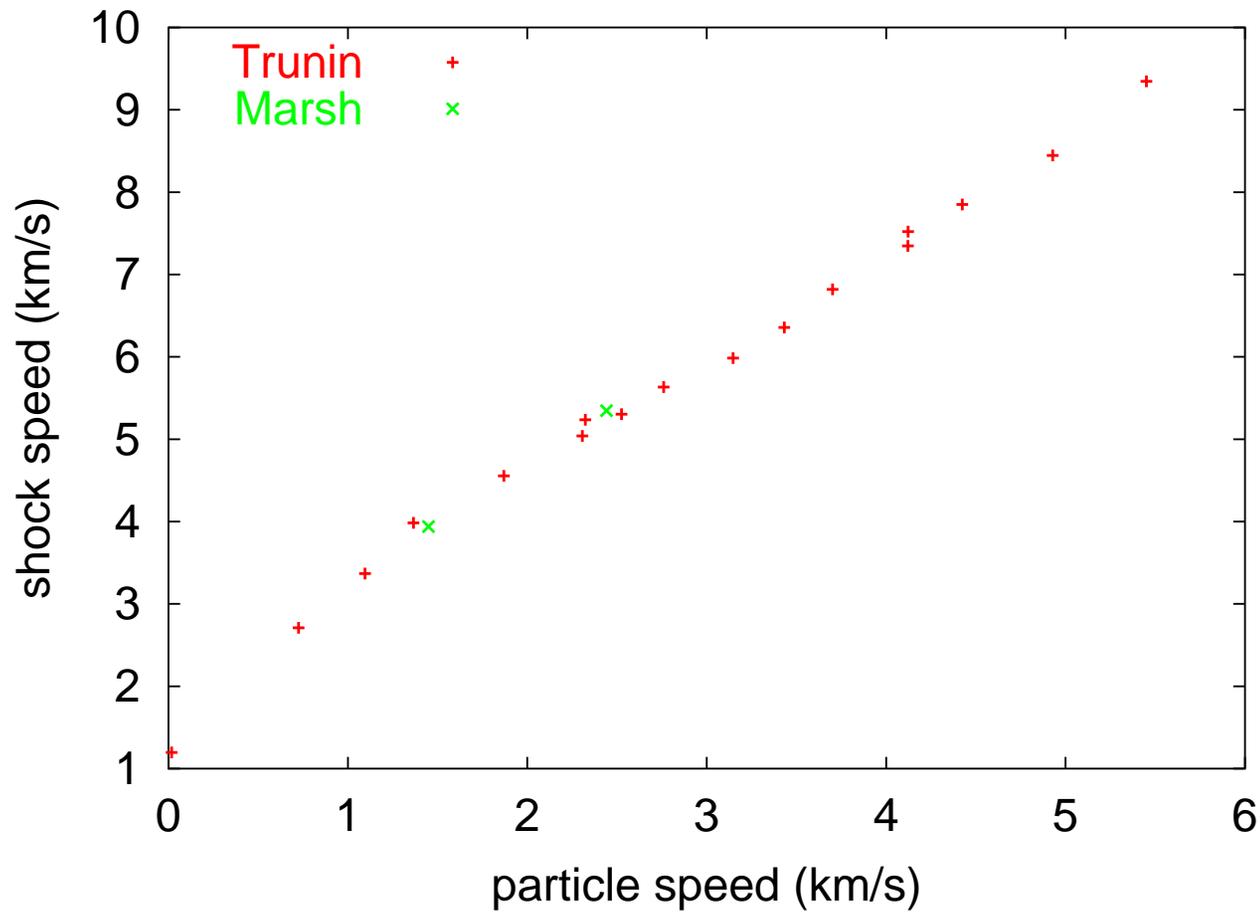
# Quasi-isentropic compression

Si, TRIDENT shot 15018:



# Shock Hugoniot data

Acetone:



# Empirical mechanical EOS

Functional fit to  $u_s - u_p$  data, e.g. Steinberg polynomial:

$$u_s = c_0 + \sum_{i=1}^n s_i \left( \frac{u_p}{u_s} \right)^{i-1} u_p,$$

R-H relations for states on the Hugoniot. Additional assumption to calculate states off-Hugoniot: Grüneisen approximation,

$$p(\rho, e) = p_{\text{ref}}(\rho) + \Gamma(\rho) [e - e_{\text{ref}}(\rho)].$$

Thus

$$p(\rho, e) = \frac{F(\rho, e)}{H^2(\rho, e)} + [\Gamma_0 + b\mu(\rho)] \rho_0 e$$

where  $\mu(\rho) = \rho/\rho_0 - 1$ ,

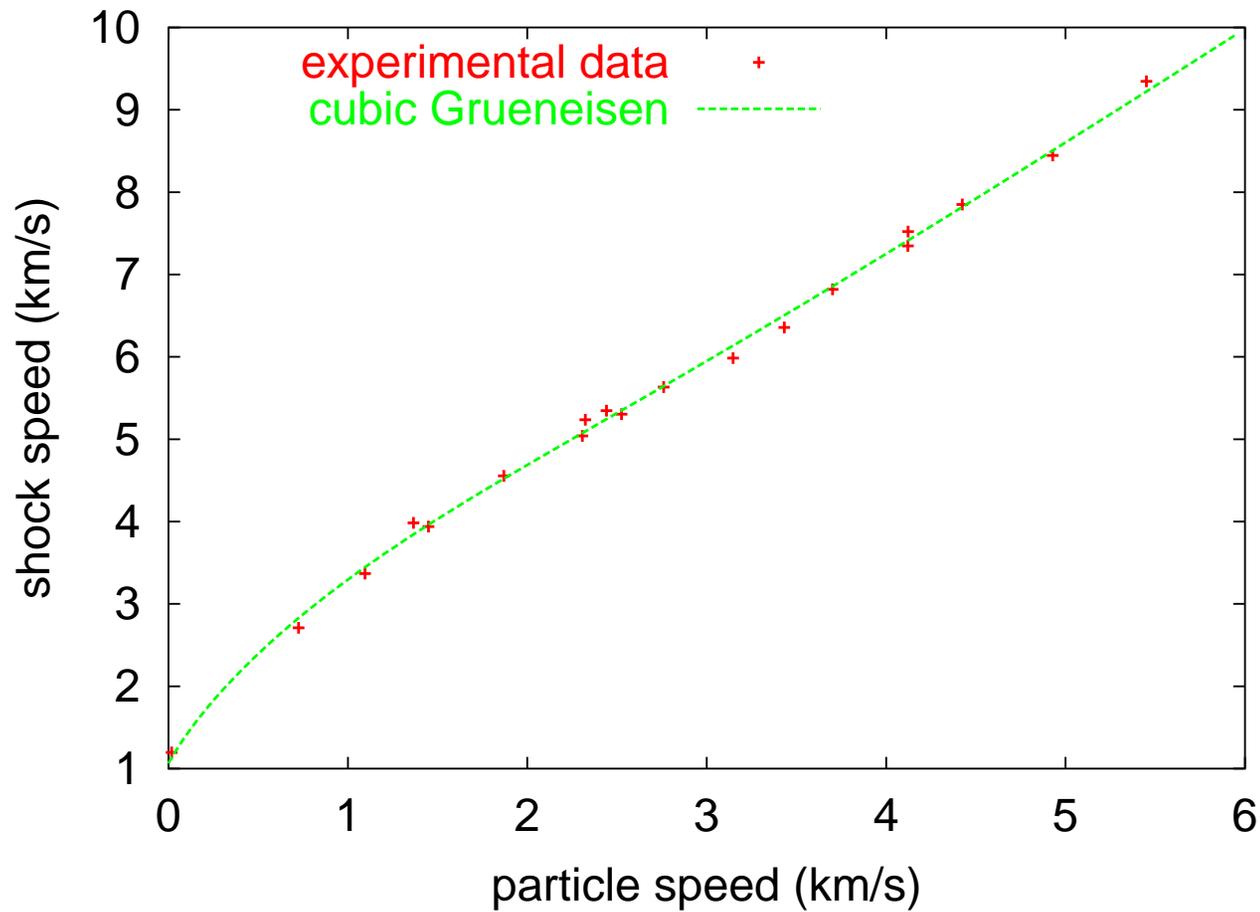
$$F(\rho, e) = \rho_0 c_0^2 \mu \{1 + \mu [(1 - \Gamma_0/2) - \mu b/2]\}$$

$$H(\rho, e) = 1 + \mu - \mu \left[ \sum_{i=1}^n s_i \left( \frac{\mu}{\mu + 1} \right)^{i-1} \right]$$

(thermodynamically incomplete).

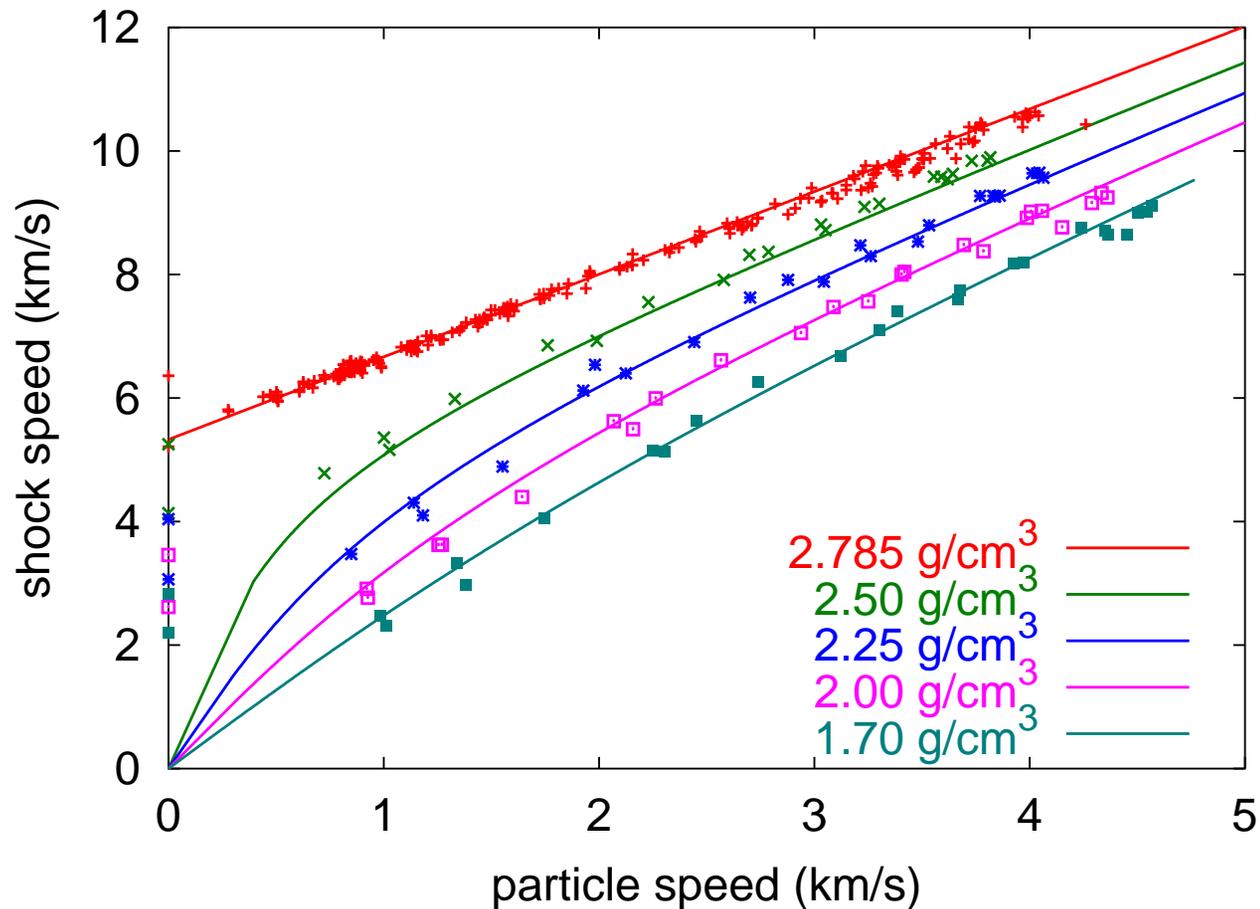
# Shock Hugoniot fit

Acetone:



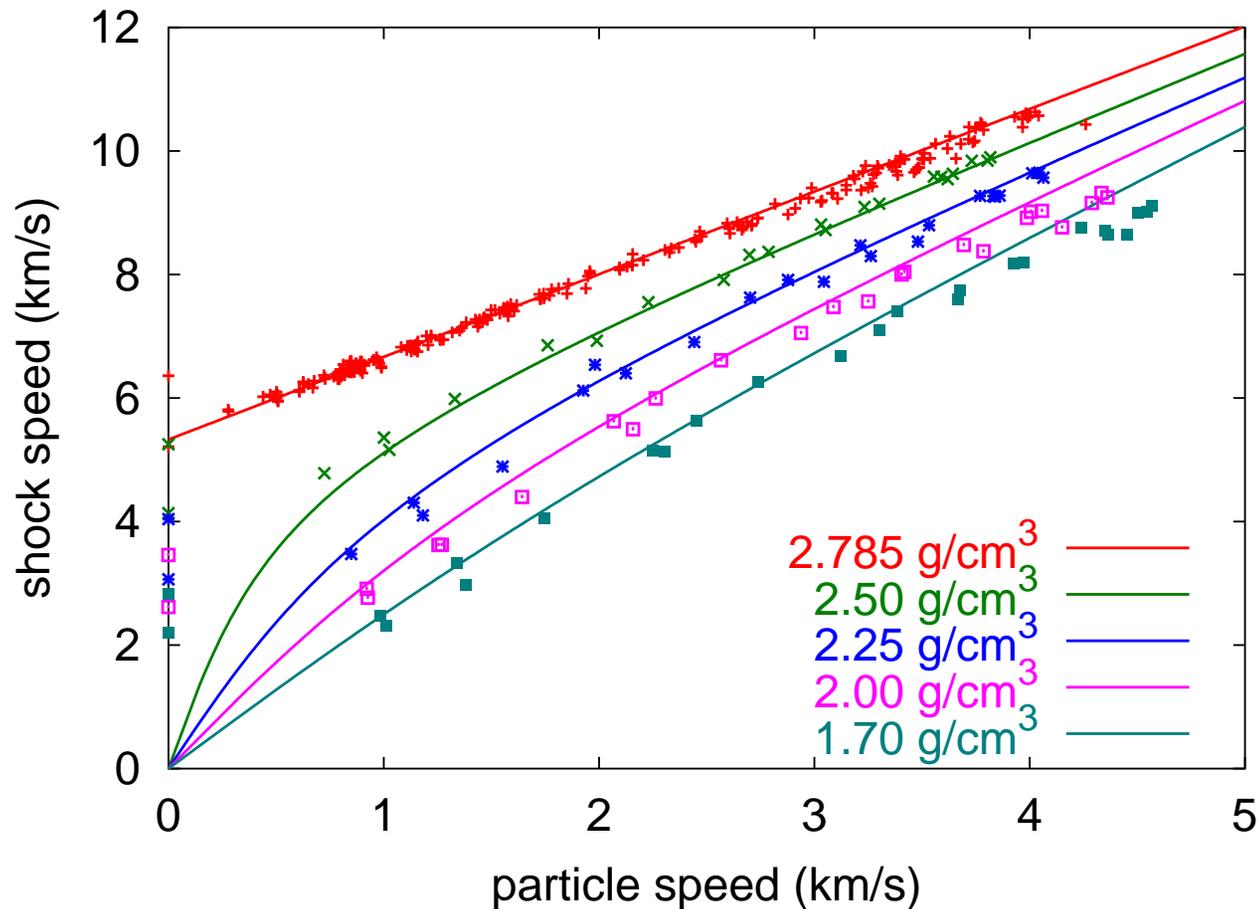
# Accuracy of empirical EOS

Shock Hugoniot for solid and porous Al; empirical EOS fitted to  $u_s - u_p$ ;  $\Gamma(\rho)$  estimated from slope of  $u_s - u_p$ :



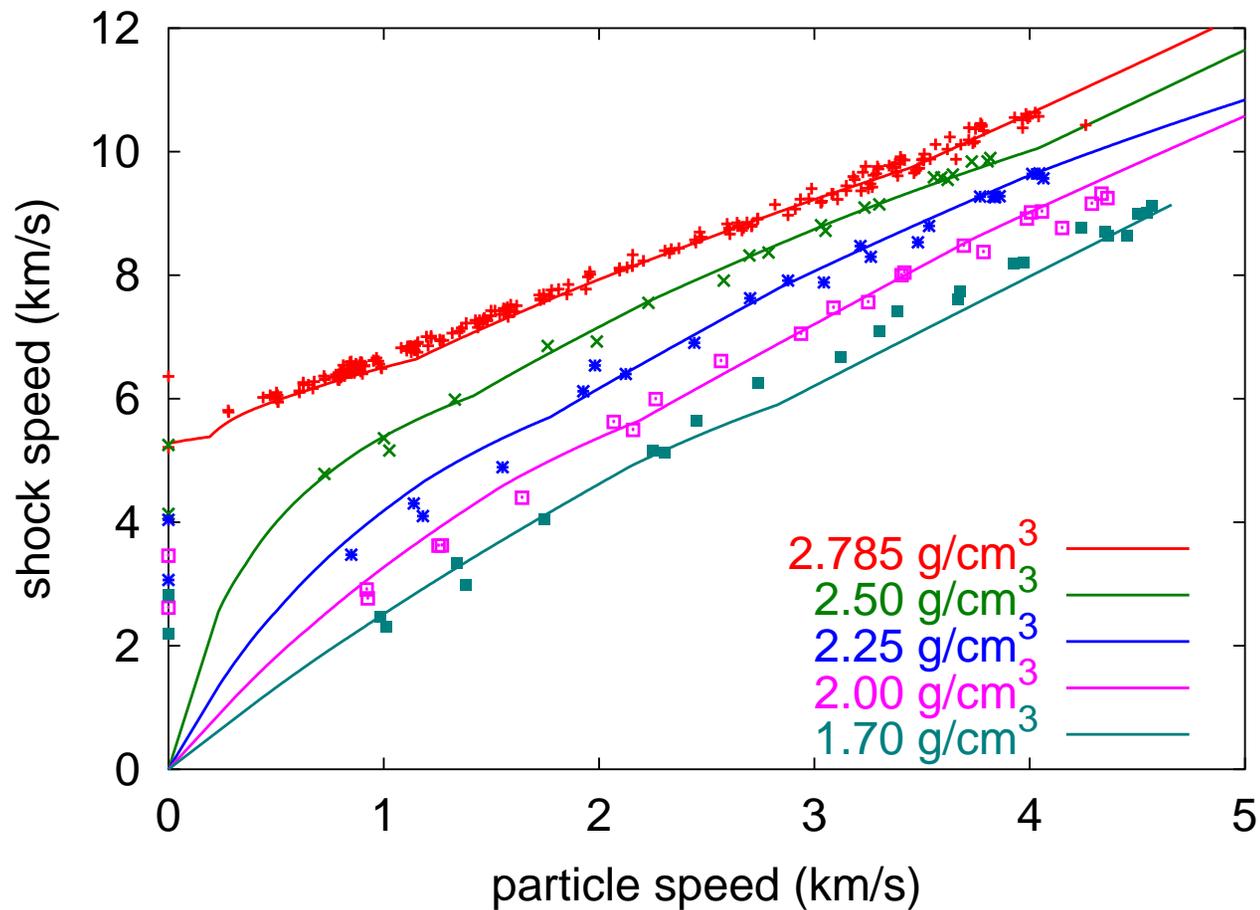
# Accuracy of empirical EOS

Shock Hugoniot for solid and porous Al; empirical EOS fitted to  $u_s - u_p$ ;  $\Gamma(\rho)$  adjusted to reproduce porous Hugoniot points:



# Accuracy of theoretical EOS

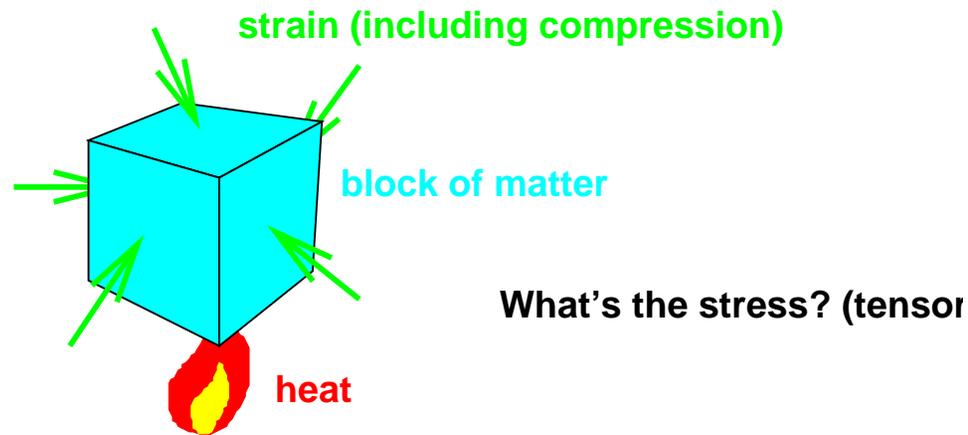
Shock Hugoniot for solid and porous Al; theoretical EOS uses *ab initio* quantum mechanics:



# Limitations and current research

- Thermodynamic equilibrium
  - electronic states: ‘non-LTE effects’ (below ns)
  - atomic configuration: phase change dynamics (ps to Myr)
- Stress tensor: decomposition into EOS and stress deviator not always valid.
- Measurement of temperatures in shock experiments is difficult – how to test theoretical EOS?
- Accuracy of quantum mechanical EOS predictions, especially for *f*-electron materials (lanthanides and actinides).

# Stress in continuum mechanics



Strain tensor  $e(\vec{r})$ , stress tensor  $\tau(\rho, T, e)$  – generalized EOS.

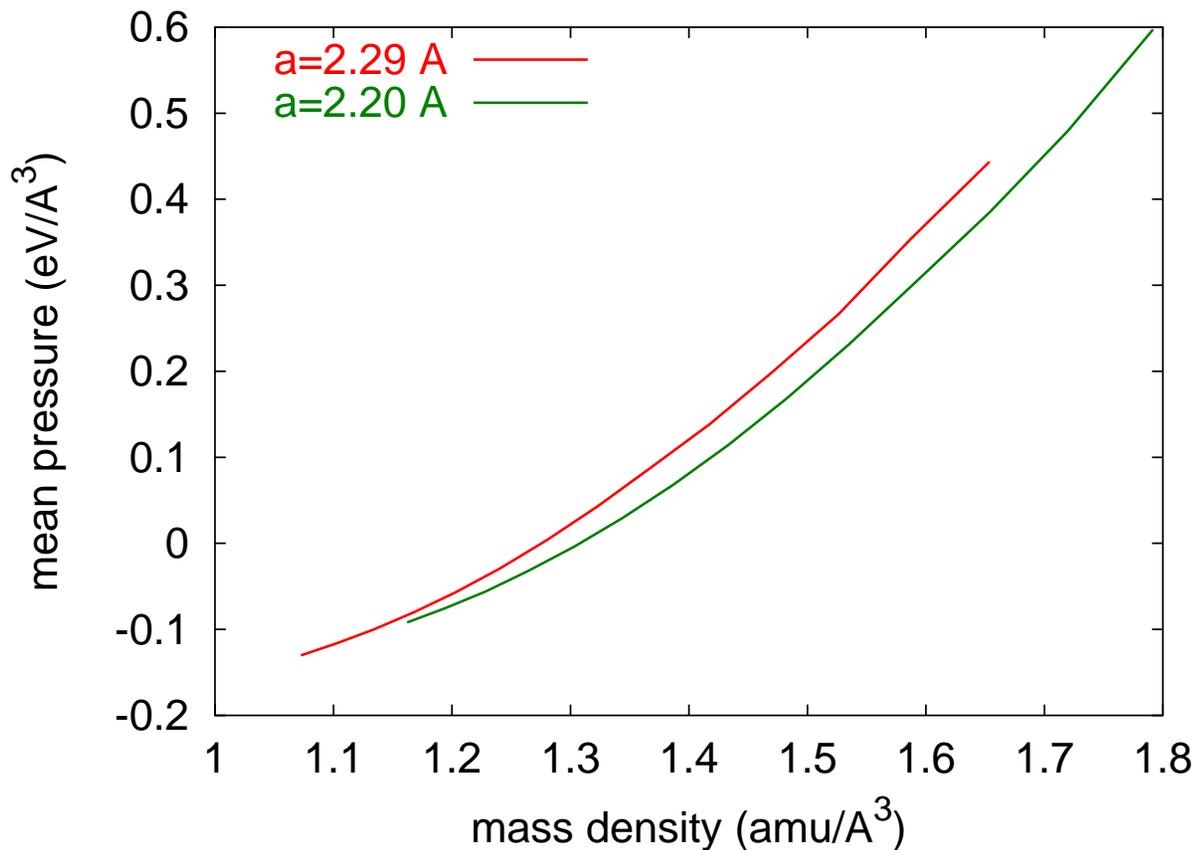
Deviatoric decomposition:

$$e = \mu I + \epsilon, \quad \tau = -pI + \sigma; \quad \mu \equiv \frac{1}{3} \text{Tr } e, \quad p \equiv -\frac{1}{3} \text{Tr } \tau$$

Assumption that scalar properties are independent:  $p(\mu, T)$ ,  $\sigma(\mu, T, \epsilon)$ .

# Uniqueness of the EOS

QM prediction of stress tensor in Be for different elastic strains:



D.C. Swift and G.J. Ackland, Appl. Phys. Lett. **86**, 6 (2003).  
– at the end of the day, the EOS may not really exist!